thermoscientific

EA-IRMS: Using isotope fingerprints to track sources of PM_{2.5} in air pollution

Authors

Christopher Brodie Thermo Fisher Scientific (Bremen) GmbH, Germany

Keywords

Air pollution, Carbon, EA IsoLink IRMS System, Haze events, Isotope fingerprints, Nitrogen, PM_{2.5}, Source apportionment, Sulfur

Goal

Trace the origin of PM_{2.5} particles in air pollution using carbon, nitrogen and sulfur isotope fingerprints.

Introduction

Air pollution, also known as haze events, is a growing global issue that results from a high concentration of sub-micron particles (PM_{2.5}) in air in the form of sulfate aerosols and nitrous oxides derived from fossil fuel burning, ozone and volatile organic compounds^{1,2,3}. PM_{2.5} particles have a high scattering coefficient on solar radiation and combined with high humidity, reduce visibility significantly. These factors impact the surface temperature of the earth, thereby affecting weather and climate, and carry serious implications for human cardiovascular and respiratory health^{1,4}. In order to improve air quality, it is important to identify the source of air pollutants (referred to as source apportionment) and application of isotope fingerprint analysis using isotope ratio mass spectrometry (IRMS) is an informative and rapid technique for achieving this^{1,2,3}.



APPLICATION BRIEF 30482

Analytical configuration

PM_{a E} particles were collected on quartz microfiber filters (254 mm x 203 mm x 2.2 µm) in Seoul, South Korea from November 2016 through January 2017, using a high volume PM₂₅ impactor sampler. Sampling times reflected changes in air pollution levels. For simultaneous nitrogen, carbon and sulfur stable isotope analysis, 5-6 mg of guartz filter was weighed into tin capsules. The tin capsules were then introduced to the combustion reactor of the Thermo Scientific[™] EA IsoLink[™] IRMS System from the Thermo Scientific[™] MAS Plus Autosampler and combusted in the presence of oxygen. The gases were then passed over hot copper where nitrous oxides were reduced to N₂ and excess oxygen was removed from the gas stream, producing N₂, CO₂ and SO, gas for isotope analysis. Analysis time was 10 minutes, using 1.4 liters of helium per sample. Each filter sample was analyzed in triplicate. Carbon and Nitrogen isotope values were referenced by single point calibration against USGS 40 and IAEA N1, respectively and sulfur isotope ratios were referenced by 3-point calibration against IAEA S1, IAEA S2 and USGS 42.

Isotope fingerprints in air pollution

For air pollution studies, carbon isotope fingerprints principally indicate a fossil fuel source, such as coal or diesel emissions from road vehicles and/or maritime vessels. The global carbon isotope fingerprint range spans -19‰ to -35‰, indicating that fossil fuel origin can be identified^{3,5}, where vehicle emissions are reported in the range of -25‰ to -28‰^{3,6}. However, marine biota and biomass burning can also contribute to the source of carbon in air particles⁶.

Besides the large nitrogen atmospheric reservoir, nitrogen in air particles/aerosols is present as NO_x , deriving from coal-fired plants, vehicle and maritime emissions and in China, winter heating via coal burning^{1,6}. The nitrogen isotope fingerprints of coal-fired power plants range between 6‰ and 20‰² with the nitrogen isotope composition of coal from China documented in the range of -6‰ to +4‰, similar to that reported from Europe and North America where the range is -2.5‰ to +3.5‰⁷. NO_x contributions from vehicle and maritime emissions have been reported to have nitrogen isotope fingerprints of -13‰ to +8‰², with a recent study in Seoul reporting values of ca. 8.6‰⁶. Contributions from soils, particularly fertilized soils, are reported in the range of -49‰ to -20‰². The sulfur isotope fingerprint in air particles and aerosols can differentiate between man-made and natural sources. Maritime sulfate aerosols, deriving from sea salt sulfate and dimethyl sulfide (DMS) as a primary productivity by-product, tend to have a sulfur isotope fingerprint of 19‰ to 21‰¹. Man-made sources of sulfate have a sulfur isotope fingerprint of +1 to +11‰⁸. Biogenic sulfur released from terrestrial sources, such as soils and wetlands, ranges from -10‰ to -2‰¹. The sulfur isotope fingerprint in coal from north China has been reported to be 6.6‰¹, however the range of sulfur in coals can vary from -15‰ to+50‰⁶, depending on the source of organic material at the time of sedimentation⁷.

Haze events in Seoul, South Korea

The nitrogen, carbon and sulfur isotope fingerprints from the samples collected on the quartz filters are presented in Table 1 and illustrated in Figure 1. For the analyzed haze events covering November 2016 through January 2017, the carbon isotope fingerprint (δ^{13} C) ranged from -26.33‰ to -27.59‰ (seasonal average: -26.8±0.5‰); the nitrogen isotope fingerprint (δ^{15} N) ranged from 4.19‰ to 8.58‰ (seasonal average: 6.7±1.5‰); and the sulfur isotope fingerprint (δ^{34} S) ranged from 4.01‰ to 7.35‰ (seasonal average: 5.8±1.2‰).



Figure 1. Nitrogen and sulfur isotope fingerprints from filtered air from Seoul, South Korea.

Table 1. $\delta^{15}N$, $\delta^{13}C$ and $\delta^{34}S$ from air filter samples as mean $\pm 1\sigma$ (n = 3).

Sample	δ ¹⁵ Ν _{AIR} (‰)	µg N	δ ¹³ C _{VPDB} (‰)	µg C	δ ³⁴ S _{VCDT} (‰)	µg S
Filter 1	6.40 ± 0.25	2.6	-26.75 ± 0.38	36.9	5.59 ± 0.33	1.1
Filter 2	7.28 ± 0.20	1.9	-27.59 ± 0.15	35.8	6.05 ± 0.37	0.5
Filter 3	7.35 ± 0.68	3.5	-27.15 ± 0.48	51.0	4.82 ± 0.09	1.1
Filter 4	8.58 ± 0.04	10.4	-26.88 ± 0.31	67.6	4.01 ± 0.15	2.3
Filter 5	6.20 ± 0.60	1.8	-26.25 ± 0.17	38.0	6.73 ± 0.10	0.9
Filter 6	4.19 ± 0.47	1.6	-26.33 ± 0.23	44.5	7.35 ± 0.18	1.1

The δ^{13} C value showed little variation and is indicative of vehicle emissions, aligning with recently reported data form Seoul⁶. The δ^{34} S values decreased from 7.35‰ to 4.01‰ and the δ^{15} N values increased from 4.2‰ to 8.6‰ as air pollution increased. The increase in δ^{15} N values and total nitrogen is principally related to coal combustion and secondarily to vehicle emissions⁶. The δ^{34} S values are indicative of a fossil fuel source and not maritime aerosols or biogenic sulfur sources, correlating with coal-burning during winter heating in northern and north-eastern China^{1,6}.

A rapid way to detect $PM_{2.5}$ sources in air pollution

The analytical approach here represents a direct measurement of PM₂₅ particles on quartz filters, utilizing the combined high sensitivity of the EA IsoLink IRMS System and fast analysis times. The analysis was complete in 10 minutes, with additional time taken to sub-sample, weigh and load samples into the autosampler for analysis. This represents a significantly shorter time from sample to data compared with longer sample chemical preparation techniques, such as, precipitating the sulfur as barium sulfate¹, which has been employed to obtain sufficient elemental sulfur concentrations for analysis. Our approach is faster, less labor intensive and lower cost without compromising data quality. In addition, whilst nitrogen analysis using the denitrification method⁶ provides a detailed source insight, direct measurement as shown here is sufficient for source identification when combined with carbon and sulfur isotope fingerprints.

Summary

The measurement of carbon, nitrogen and sulfur isotope fingerprints of $PM_{2.5}$ particles on quartz filter papers allows source identification. Of the samples analyzed here, $PM_{2.5}$ particles are generated from fossil fuel sources derived from coal combustion due to winter heating and from vehicle and maritime emissions. Combining these direct measurements from quartz filters with isotope fingerprints reported in the literature, source identification of $PM_{2.5}$ particles is possible.

Using the EA IsoLink IRMS System for source identification of $PM_{2.5}$ particles in air pollution, laboratories provides:

- A rapid analytical technique for source identification using isotope fingerprints
- An efficient, automated technique for source identification that *does not require* sample chemical clean up, with its associated high labour and consumable costs
- A reliable analytical technique that produces high quality data comparable with published literature^{1,2,3,6}

References

- 1. Han, X., et al. (2016) Using Stable Isotopes to trace sources and formation processes of sulfate aerosols from Beijing, China. Nature Scientific Reports. 6: 29958.
- Beyn, F., et al. (2015) Do N-isotopes in atmospheric nitrate deposition reflect air pollution levels? Atmos. Environ., 107, 281-288.
- Dai et al (2015) Chemical and stable carbon isotopic composition of PM2:5 from on-road vehicle emissions in the PRD region and implications for vehicle emission control policy. Atmos. Chem. Phys., 15, 3097-3108.
- IPCC. Climate Change 2007: The Physical Science Basis. In: Solomons, S. *et al.* (Eds), *Contribution of Working Group 1 to the Fourth Assessment Report of the Intergovernmental Panel on Climate Change*. Cambridge University Press, Cambridge (2007).
- Masalaite, A., Garbaras, A., Remeikis, V. (2012) Stable Isotopes in environmental Investigations. Lith. J. Phys. 52, 261-268.
- 6. Park, Y-M., Park, K-S., Kim, H., Yu, S-M., Kim, M-S., Kim, JOY, Ahn, J-Y., Lee, M-D., Seok, K-S., Kim, Y-H. (2018) Characterizing isotopic compositions of TC-C, $NO_3^{-}N$, and $NH_4^{+}-N$ in PM_{2.5} in South Korea: Impact of China's winter heating. Environ. Poll. 233, 733-744.
- Xiao, H-Y., Liu, C-Q. (2011) The elemental and isotopic composition of sulfur and nitrogen in Chinese coals. Org. Geochem. 42, 84-93.
- Shaheen, R. et al. (2014) Large sulfur-isotope anomaly in nonvolcanic sulfate aerosol and its implication for the Archean atmosphere. Proc. Natl. Acad. Sci. 111, 11979-11983.

Find out more at thermofisher.com/lsotopeFingerprints

©2018 Thermo Fisher Scientific Inc. All rights reserved. All trademarks are the property of Thermo Fisher Scientific. This information is presented as an example of the capabilities of Thermo Fisher Scientific products. It is not intended to encourage use of these products in any manner that might infringe the intellectual property rights of others. Specifications, terms and pricing are subject to change. Not all products are available in all countries. Please consult your local sales representative for details. AB30482-EN 0118

